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***N*-[4-(*p*-Toluenesulfonamido)phenyl]sulfonylacetamide**

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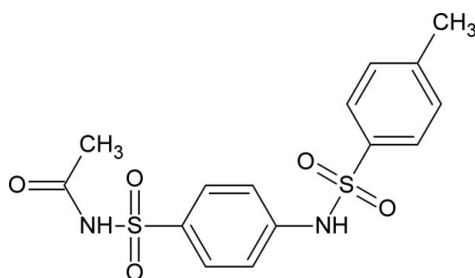
Received 18 December 2009; accepted 29 December 2009

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.038; wR factor = 0.114; data-to-parameter ratio = 16.5.

In the title compound, $\text{C}_{15}\text{H}_{16}\text{N}_2\text{O}_5\text{S}_2$, the dihedral between the two aromatic rings is $81.33(6)^\circ$. In the crystal, pairs of $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules into centrosymmetric dimers, which are further connected via $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds into a chain running along $[\bar{1}01]$.

Related literature

For the synthesis and biological activity of the title compound, see: Deng & Mani (2006). For a related structure, see: Ashfaq *et al.* (2009).



Experimental

Crystal data

$\text{C}_{15}\text{H}_{16}\text{N}_2\text{O}_5\text{S}_2$
 $M_r = 368.42$

Monoclinic, $P2_1/n$
 $a = 9.8077(4)$ Å

$b = 10.0782(4)$ Å
 $c = 17.3081(7)$ Å
 $\beta = 100.290(2)^\circ$
 $V = 1683.28(12)$ Å³
 $Z = 4$

Mo $K\alpha$ radiation
 $\mu = 0.34$ mm⁻¹
 $T = 296$ K
 $0.48 \times 0.14 \times 0.05$ mm

Data collection

Bruker Kappa APEXII CCD diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2007)
 $T_{\min} = 0.852$, $T_{\max} = 0.981$

16294 measured reflections
 3715 independent reflections
 2787 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.032$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.114$
 $S = 1.02$
 3715 reflections
 225 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.34$ e Å⁻³
 $\Delta\rho_{\min} = -0.31$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1N}\cdots\text{O5}^{\text{i}}$	0.81 (2)	2.06 (2)	2.848 (2)	162 (2)
$\text{N2}-\text{H2N}\cdots\text{O1}^{\text{ii}}$	0.82 (2)	2.15 (2)	2.950 (2)	167 (2)

Symmetry codes: (i) $-x + 2, -y + 2, -z + 2$; (ii) $x + \frac{1}{2}, -y + \frac{3}{2}, z - \frac{1}{2}$.

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and PLATON (Spek, 2009); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5152).

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supplementary materials

Acta Cryst. (2010). E66, o299 [doi:10.1107/S1600536809055706]

N-[4-(*p*-Toluenesulfonamido)phenylsulfonyl]acetamide

M. Ashfaq, I. U. Khan, M. N. Arshad, H. Ahmad and M. N. Asghar

Comment

The bond angles and length are in comparison with the previously published crystal structure of *N*-Acetyl-4-(benzenesulfonamido)benzenesulfonamide (II) (Ashfaq *et al.*, 2009). The dihedral angle between the two aromatic rings is 81.33(0.06)°, the acetamido group is oriented at 79.13(0.11)° and 14.42(0.26)° with respect to the central aromatic ring (C7/C8/C9/C10/C11/C12) and toluene ring. The compound may be stabilized by the formation of N–H···O type hydrogen bondings. The acetamido N–H interact with oxygen of C=O moiety and forms a $R_2^2(20)$ ring. The hydrogen bonding interaction between the sulfonamido N–H and SO₂ gives rise in the formation of infinite long chain along [-1 0 1] (Fig. 2 Table, 1).

Experimental

The title compound was prepared using a literature method (Deng & Mani, 2006). Sodium sulphacetamide (2 g, 9.3 mmol) was dissolved in distilled water, and then toluene sulfonyl chloride (1.77 g, 9.3 mmol) was added with stirring at room temperature. The pH was maintained at 8-9, strictly using Na₂CO₃ (1 M). The completion of reaction was observed by the consumption of the suspended toluene sulfonyl chloride. On completion, pH was adjusted to 2-3 using HCl (2 N). The precipitate formed was filtered, washed with distilled water and recrystallized from methanol.

Refinement

The H-atoms bonded to C were positioned geometrically and refined using a riding model with C–H = 0.93 Å, $U(H) = 1.2 U_{eq}(C)$ for aromatic and C–H = 0.96 Å for CH₃, $U(H) = 1.5 U_{eq}(C)$ for CH₃. The N-H H atoms were located in difference map and their coordinates were refined with $U(H) = 1.2 U_{eq}(N)$ for N atoms.

Figures

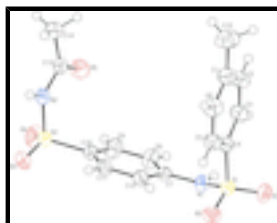


Fig. 1. The structure of the title compound with atomic label and displacement ellipsoids drawn at the 50% probability level.

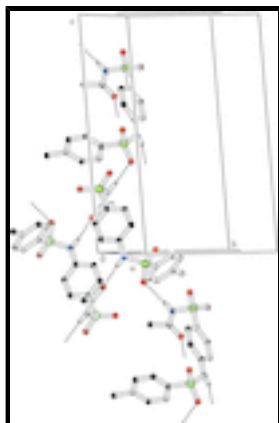


Fig. 2. Packing diagram for the title compound with hydrogen bonding drawn as dashed lines. Hydrogen atoms not involve in hydrogen bonding have been omitted.

***N*-[4-(*p*-Toluenesulfonamido)phenylsulfonyl]acetamide**

Crystal data

$C_{15}H_{16}N_2O_5S_2$

$M_r = 368.42$

Monoclinic, $P2_1/n$

Hall symbol: $-P\ 2_1n$

$a = 9.8077\ (4)\ \text{\AA}$

$b = 10.0782\ (4)\ \text{\AA}$

$c = 17.3081\ (7)\ \text{\AA}$

$\beta = 100.290\ (2)^\circ$

$V = 1683.28\ (12)\ \text{\AA}^3$

$Z = 4$

$F(000) = 768$

$D_x = 1.454\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 4606 reflections

$\theta = 2.4\text{--}26.8^\circ$

$\mu = 0.34\ \text{mm}^{-1}$

$T = 296\ \text{K}$

Needle, red

$0.48 \times 0.14 \times 0.05\ \text{mm}$

Data collection

Bruker Kappa APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube
graphite

φ and ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2007)

$T_{\min} = 0.852$, $T_{\max} = 0.981$

16294 measured reflections

3715 independent reflections

2787 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.032$

$\theta_{\max} = 27.1^\circ$, $\theta_{\min} = 2.4^\circ$

$h = -11 \rightarrow 12$

$k = -12 \rightarrow 12$

$l = -22 \rightarrow 21$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.038$

Primary atom site location: structure-invariant direct
methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring
sites

$wR(F^2) = 0.114$

$S = 1.02$

3715 reflections

225 parameters

0 restraints

H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0603P)^2 + 0.5119P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} = 0.001$$

$$\Delta\rho_{\max} = 0.34 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.31 \text{ e } \text{\AA}^{-3}$$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\text{sigma}(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.58372 (5)	0.79324 (5)	1.04610 (3)	0.03142 (15)
S2	0.91252 (5)	0.97819 (6)	0.73350 (3)	0.03660 (16)
O1	0.55834 (15)	0.82933 (16)	1.12253 (9)	0.0429 (4)
O2	0.47119 (14)	0.75141 (16)	0.98763 (9)	0.0422 (4)
O3	0.82154 (16)	0.9285 (2)	0.66630 (9)	0.0553 (5)
O4	0.96901 (18)	1.10836 (16)	0.73294 (11)	0.0532 (5)
O5	1.16225 (16)	0.96040 (18)	0.85525 (9)	0.0501 (4)
N1	0.65425 (17)	0.92526 (18)	1.01581 (10)	0.0315 (4)
H1N	0.693 (2)	0.971 (2)	1.0518 (14)	0.038*
N2	1.04204 (18)	0.86957 (19)	0.74560 (10)	0.0341 (4)
H2N	1.034 (2)	0.811 (2)	0.7122 (14)	0.041*
C1	0.9121 (2)	0.4715 (2)	1.08120 (15)	0.0427 (5)
C2	0.9326 (2)	0.5914 (2)	1.12183 (15)	0.0470 (6)
H2	1.0146	0.6052	1.1572	0.056*
C3	0.8338 (2)	0.6893 (2)	1.11050 (13)	0.0413 (5)
H3	0.8485	0.7686	1.1382	0.050*
C4	0.7115 (2)	0.6690 (2)	1.05720 (12)	0.0316 (4)
C5	0.6907 (2)	0.5522 (2)	1.01500 (13)	0.0382 (5)
H5	0.6100	0.5395	0.9784	0.046*
C6	0.7907 (2)	0.4546 (2)	1.02772 (15)	0.0436 (5)
H6	0.7761	0.3756	0.9997	0.052*
C7	0.71121 (19)	0.93381 (19)	0.94715 (11)	0.0278 (4)
C8	0.8064 (2)	1.0347 (2)	0.94356 (13)	0.0404 (5)
H8	0.8291	1.0927	0.9856	0.048*
C9	0.8672 (2)	1.0495 (2)	0.87847 (14)	0.0422 (5)

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H9	0.9305	1.1176	0.8764	0.051*
C10	0.83393 (19)	0.9629 (2)	0.81628 (12)	0.0306 (4)
C11	0.7379 (2)	0.8629 (2)	0.81853 (12)	0.0330 (5)
H11	0.7152	0.8055	0.7761	0.040*
C12	0.6760 (2)	0.8483 (2)	0.88342 (12)	0.0326 (5)
H12	0.6110	0.7815	0.8847	0.039*
C13	1.1547 (2)	0.8753 (2)	0.80521 (12)	0.0354 (5)
C14	1.2619 (3)	0.7714 (3)	0.80379 (15)	0.0567 (7)
H14A	1.3395	0.8090	0.7847	0.085*
H14B	1.2234	0.7003	0.7698	0.085*
H14C	1.2921	0.7377	0.8559	0.085*
C15	1.0205 (3)	0.3644 (3)	1.09553 (19)	0.0622 (8)
H15A	1.1045	0.3959	1.0806	0.093*
H15B	0.9882	0.2876	1.0649	0.093*
H15C	1.0381	0.3414	1.1502	0.093*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0265 (2)	0.0369 (3)	0.0317 (3)	-0.0007 (2)	0.00741 (19)	0.0050 (2)
S2	0.0365 (3)	0.0453 (3)	0.0291 (3)	0.0072 (2)	0.0089 (2)	0.0104 (2)
O1	0.0458 (9)	0.0506 (10)	0.0363 (9)	0.0046 (7)	0.0186 (7)	0.0054 (7)
O2	0.0288 (7)	0.0490 (9)	0.0467 (9)	-0.0064 (7)	0.0012 (7)	0.0068 (7)
O3	0.0435 (9)	0.0934 (14)	0.0263 (8)	0.0096 (9)	-0.0005 (7)	0.0080 (9)
O4	0.0615 (11)	0.0412 (9)	0.0638 (11)	0.0059 (8)	0.0301 (9)	0.0200 (9)
O5	0.0470 (9)	0.0621 (11)	0.0386 (9)	-0.0020 (8)	0.0001 (7)	-0.0165 (8)
N1	0.0343 (9)	0.0320 (10)	0.0286 (9)	-0.0029 (7)	0.0069 (7)	-0.0014 (7)
N2	0.0366 (9)	0.0396 (11)	0.0259 (9)	0.0044 (8)	0.0051 (7)	-0.0057 (8)
C1	0.0388 (12)	0.0384 (13)	0.0519 (14)	0.0012 (10)	0.0107 (10)	0.0152 (11)
C2	0.0344 (11)	0.0510 (15)	0.0502 (15)	0.0006 (10)	-0.0070 (10)	0.0055 (12)
C3	0.0388 (12)	0.0413 (13)	0.0412 (13)	-0.0034 (10)	0.0004 (9)	-0.0022 (10)
C4	0.0293 (10)	0.0340 (11)	0.0314 (11)	-0.0024 (8)	0.0054 (8)	0.0043 (9)
C5	0.0334 (11)	0.0390 (12)	0.0408 (12)	-0.0047 (9)	0.0024 (9)	0.0011 (10)
C6	0.0458 (13)	0.0312 (12)	0.0541 (15)	-0.0019 (10)	0.0095 (11)	-0.0002 (11)
C7	0.0261 (9)	0.0279 (10)	0.0294 (10)	0.0036 (8)	0.0048 (7)	0.0034 (8)
C8	0.0468 (12)	0.0374 (12)	0.0399 (13)	-0.0130 (10)	0.0157 (10)	-0.0106 (10)
C9	0.0472 (12)	0.0362 (12)	0.0474 (14)	-0.0133 (10)	0.0203 (10)	-0.0059 (10)
C10	0.0299 (10)	0.0337 (11)	0.0290 (10)	0.0040 (8)	0.0077 (8)	0.0049 (9)
C11	0.0343 (10)	0.0363 (12)	0.0270 (10)	-0.0003 (9)	0.0014 (8)	-0.0019 (9)
C12	0.0298 (10)	0.0345 (11)	0.0328 (11)	-0.0069 (8)	0.0035 (8)	-0.0002 (9)
C13	0.0328 (10)	0.0478 (13)	0.0260 (11)	0.0016 (9)	0.0069 (8)	0.0010 (10)
C14	0.0453 (14)	0.080 (2)	0.0434 (14)	0.0217 (13)	0.0050 (11)	-0.0013 (13)
C15	0.0514 (15)	0.0450 (15)	0.092 (2)	0.0109 (12)	0.0188 (14)	0.0226 (15)

Geometric parameters (\AA , $^\circ$)

S1—O2	1.4215 (15)	C5—C6	1.379 (3)
S1—O1	1.4359 (15)	C5—H5	0.9300
S1—N1	1.6289 (18)	C6—H6	0.9300

S1—C4	1.758 (2)	C7—C8	1.389 (3)
S2—O3	1.4244 (17)	C7—C12	1.394 (3)
S2—O4	1.4247 (18)	C8—C9	1.374 (3)
S2—N2	1.6615 (18)	C8—H8	0.9300
S2—C10	1.751 (2)	C9—C10	1.379 (3)
O5—C13	1.211 (3)	C9—H9	0.9300
N1—C7	1.403 (2)	C10—C11	1.385 (3)
N1—H1N	0.81 (2)	C11—C12	1.377 (3)
N2—C13	1.372 (3)	C11—H11	0.9300
N2—H2N	0.82 (2)	C12—H12	0.9300
C1—C6	1.382 (3)	C13—C14	1.488 (3)
C1—C2	1.394 (4)	C14—H14A	0.9600
C1—C15	1.504 (3)	C14—H14B	0.9600
C2—C3	1.372 (3)	C14—H14C	0.9600
C2—H2	0.9300	C15—H15A	0.9600
C3—C4	1.392 (3)	C15—H15B	0.9600
C3—H3	0.9300	C15—H15C	0.9600
C4—C5	1.381 (3)		
O2—S1—O1	119.38 (9)	C1—C6—H6	119.3
O2—S1—N1	109.48 (9)	C8—C7—C12	119.38 (18)
O1—S1—N1	104.07 (9)	C8—C7—N1	117.08 (18)
O2—S1—C4	108.32 (10)	C12—C7—N1	123.53 (18)
O1—S1—C4	108.50 (9)	C9—C8—C7	120.7 (2)
N1—S1—C4	106.35 (9)	C9—C8—H8	119.7
O3—S2—O4	120.38 (11)	C7—C8—H8	119.7
O3—S2—N2	102.99 (10)	C8—C9—C10	119.6 (2)
O4—S2—N2	108.54 (10)	C8—C9—H9	120.2
O3—S2—C10	109.52 (10)	C10—C9—H9	120.2
O4—S2—C10	108.37 (10)	C9—C10—C11	120.42 (19)
N2—S2—C10	106.09 (9)	C9—C10—S2	120.36 (16)
C7—N1—S1	125.38 (15)	C11—C10—S2	119.22 (16)
C7—N1—H1N	114.3 (17)	C12—C11—C10	120.14 (19)
S1—N1—H1N	112.5 (17)	C12—C11—H11	119.9
C13—N2—S2	124.19 (16)	C10—C11—H11	119.9
C13—N2—H2N	121.9 (17)	C11—C12—C7	119.76 (19)
S2—N2—H2N	113.9 (17)	C11—C12—H12	120.1
C6—C1—C2	118.3 (2)	C7—C12—H12	120.1
C6—C1—C15	121.4 (2)	O5—C13—N2	120.4 (2)
C2—C1—C15	120.3 (2)	O5—C13—C14	123.8 (2)
C3—C2—C1	121.1 (2)	N2—C13—C14	115.8 (2)
C3—C2—H2	119.5	C13—C14—H14A	109.5
C1—C2—H2	119.5	C13—C14—H14B	109.5
C2—C3—C4	119.5 (2)	H14A—C14—H14B	109.5
C2—C3—H3	120.2	C13—C14—H14C	109.5
C4—C3—H3	120.2	H14A—C14—H14C	109.5
C5—C4—C3	120.3 (2)	H14B—C14—H14C	109.5
C5—C4—S1	120.96 (16)	C1—C15—H15A	109.5
C3—C4—S1	118.77 (17)	C1—C15—H15B	109.5
C6—C5—C4	119.3 (2)	H15A—C15—H15B	109.5

supplementary materials

C6—C5—H5	120.3	C1—C15—H15C	109.5
C4—C5—H5	120.3	H15A—C15—H15C	109.5
C5—C6—C1	121.5 (2)	H15B—C15—H15C	109.5
C5—C6—H6	119.3		
O2—S1—N1—C7	-58.48 (18)	C15—C1—C6—C5	179.4 (2)
O1—S1—N1—C7	172.83 (16)	S1—N1—C7—C8	-158.64 (16)
C4—S1—N1—C7	58.34 (18)	S1—N1—C7—C12	22.1 (3)
O3—S2—N2—C13	-178.58 (18)	C12—C7—C8—C9	-0.9 (3)
O4—S2—N2—C13	52.8 (2)	N1—C7—C8—C9	179.8 (2)
C10—S2—N2—C13	-63.5 (2)	C7—C8—C9—C10	-0.3 (4)
C6—C1—C2—C3	1.4 (4)	C8—C9—C10—C11	1.2 (3)
C15—C1—C2—C3	-178.9 (2)	C8—C9—C10—S2	-178.87 (18)
C1—C2—C3—C4	-0.4 (4)	O3—S2—C10—C9	-151.90 (18)
C2—C3—C4—C5	-1.1 (3)	O4—S2—C10—C9	-18.8 (2)
C2—C3—C4—S1	177.79 (18)	N2—S2—C10—C9	97.58 (19)
O2—S1—C4—C5	-2.5 (2)	O3—S2—C10—C11	28.04 (19)
O1—S1—C4—C5	128.50 (18)	O4—S2—C10—C11	161.13 (16)
N1—S1—C4—C5	-120.07 (18)	N2—S2—C10—C11	-82.47 (17)
O2—S1—C4—C3	178.62 (17)	C9—C10—C11—C12	-0.8 (3)
O1—S1—C4—C3	-50.4 (2)	S2—C10—C11—C12	179.30 (16)
N1—S1—C4—C3	61.02 (19)	C10—C11—C12—C7	-0.5 (3)
C3—C4—C5—C6	1.7 (3)	C8—C7—C12—C11	1.3 (3)
S1—C4—C5—C6	-177.20 (17)	N1—C7—C12—C11	-179.41 (18)
C4—C5—C6—C1	-0.7 (3)	S2—N2—C13—O5	2.9 (3)
C2—C1—C6—C5	-0.8 (4)	S2—N2—C13—C14	-177.73 (17)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1N \cdots O5 ⁱ	0.81 (2)	2.06 (2)	2.848 (2)	162 (2)
N2—H2N \cdots O1 ⁱⁱ	0.82 (2)	2.15 (2)	2.950 (2)	167 (2)

Symmetry codes: (i) $-x+2, -y+2, -z+2$; (ii) $x+1/2, -y+3/2, z-1/2$.

Fig. 1

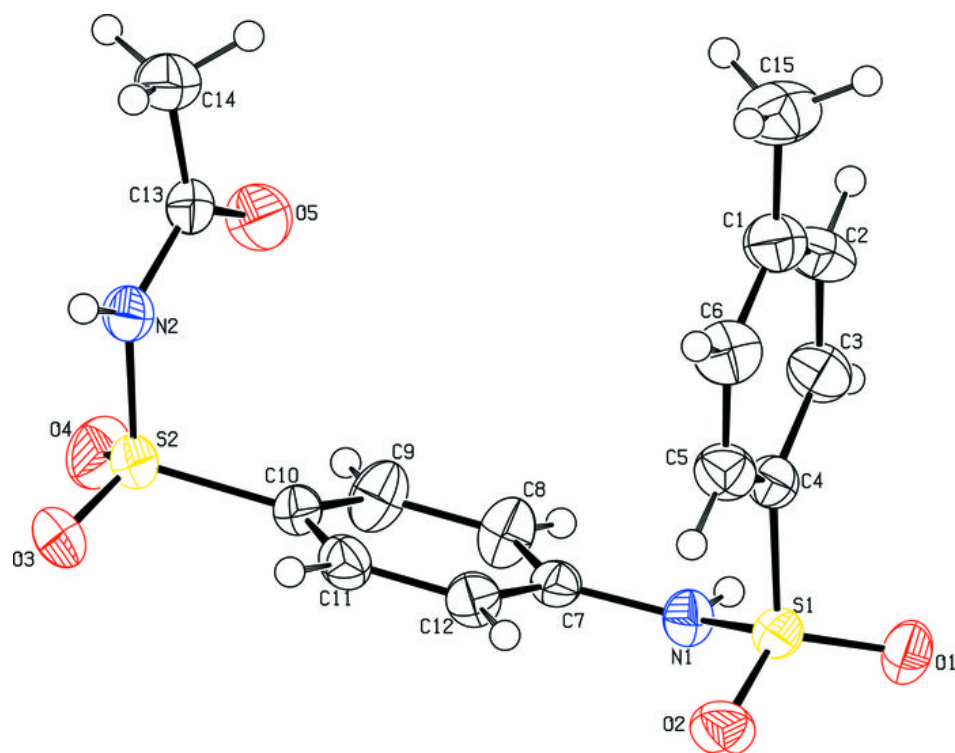


Fig. 2

